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Li-Jun Xiao and Da-Qi Wang*

College of Chemistry and Chemical Engineering, Liaocheng University, Shandong 252059, People's Republic of China

Correspondence e-mail: wdq4869@163.com

Key indicators

Single-crystal X-ray study T = 298 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.041wR factor = 0.122 Data-to-parameter ratio = 9.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

N,N'-Bis(4-dimethylaminobenzylidene)ethane-1,2-diamine

The title compound, C₂₀H₂₆N₄, was prepared by a condensation reaction between p-dimethylaminobenzaldehyde and ethylenediamine. The molecule contains a crystallographically imposed twofold axis of symmetry.

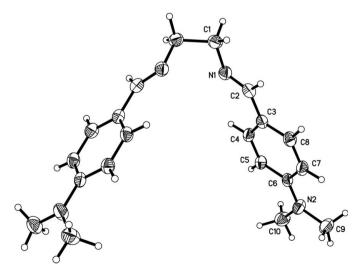
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Comment

Schiff bases are used extensively as ligands in the field of coordination chemistry (Yamada, 1999). The molecular structure of the title compound, (I), is shown in Fig. 1.

$$H_2C$$
 $N = C$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3
 CH_3

The two arms of the Schiff base are each essentially planar and are related by a crystallographically imposed twofold axis of symmetry. The bond parameters within the imine groups are comparable with those reported in the literature (Huang & Xu, 1989; Mirkhani et al., 2004).



The structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme for the asymmetric unit. Unlabelled atoms are related to labelled atoms by -x, y, $-\frac{1}{2} - z$.

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Experimental

The title compound was prepared by the reaction of p-dimethylaminobenzaldehyde and ethylenediamine (molar ratio 2:1) in absolute ethanol. The mixture was heated under reflux with stirring for 10 h, and the solvent was evaporated in a rotary vacuum evaporator. The resulting solution was filtered and the filtrate allowed to stand in air for about three weeks, after which large yellow block-shaped crystals suitable for X-ray analysis were formed. Elemental analysis found: C 74.54, H 8.17, N 17.29%; calculated for $C_{20}H_{26}N_4$: C 74.50, H 8.13, N 17.37%.

Crystal data

$C_{20}H_{26}N_4$	$D_x = 1.200 \text{ Mg m}^{-3}$
$M_r = 322.44$	Mo $K\alpha$ radiation
Monoclinic, C2/c	Cell parameters from 1166
a = 24.218 (16) Å	reflections
b = 6.534 (4) Å	$\theta = 3.3-25.6^{\circ}$
c = 12.433 (8) Å	$\mu = 0.07 \text{ mm}^{-1}$
$\beta = 114.872 \ (10)^{\circ}$	T = 298 (2) K
$V = 1785 (2) \text{ Å}^3$	Block, yellow
Z = 4	$0.37 \times 0.13 \times 0.08 \text{ mm}$

Data collection

Siemens SMART CCD area-	1577 independent reflections
detector diffractometer	990 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.030$
Absorption correction: multi-scan	$\theta_{ m max} = 25.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -20 \rightarrow 28$
$T_{\min} = 0.974, T_{\max} = 0.994$	$k = -7 \rightarrow 7$
4438 measured reflections	$l = -14 \rightarrow 14$

Refinement

$w = 1/[\sigma^2(F_0^2) + (0.0566P)^2]$
+ 0.44P]
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\text{max}} = 0.16 \text{ e Å}^{-3}$
$\Delta \rho_{\min} = -0.12 \text{ e Å}^{-3}$

Table 1Selected geometric parameters (Å, °).

N1-C2	1.258 (2)	N2-C10	1.426 (3)
N1-C1	1.454 (3)	N2-C9	1.438 (3)
N2-C6	1.375 (3)		
C2-N1-C1	116.41 (18)	C10-N2-C9	117.6 (2)
C6-N2-C10	121.17 (18)	N1-C2-C3	125.7 (2)
C6-N2-C9	120.6 (2)	N1-C2-H3	120.1 (11)

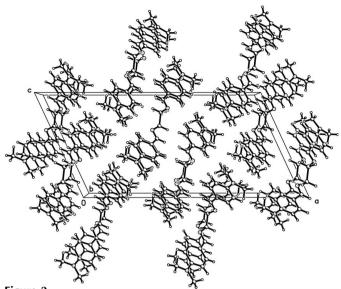


Figure 2
The crystal packing of the title compound.

All H atoms were located in a difference Fourier synthesis and refined isotropically.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996; data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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References

Huang, L. & Xu, G. X. (1989). Chin. J. Struct. Chem. 8, 1–3.
Mirkhani, V., Harkema, S. & Kia, R. (2004). Acta Cryst. C60, m343–m344.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997a). SHELXL97 and SHELXS97. University of Göttingen, Germany.

Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.

Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Yamada, S. (1999). Coord. Chem. Rev. 192, 537-555.